

in a comparison gas which is 2% O₂/ 98% N₂. All of the signals were generated with 10 V across the materials, unless otherwise specified. Blank spaces indicate that there was no detectable signal when that gas composition was contacted with that material. Unless otherwise specified, the gases were measured at 2000 ppm in N₂.

Table 5b
Change in temperature in °C

	AlVO ₄	BaCuO _{2.5}	Zn ₄ TiO ₆
CO in N ₂	18.1	-6	6
N ₂	18.1	-6	6

Example 6

This example illustrates the use of the AC impedance technique for the measurement of the response of 19 metal oxide semiconducting materials in the presence of 4 gas compositions at 400°C. The signals listed in Table 6 below are the ratios of the magnitudes of the impedances of the materials when exposed to the gas compositions shown to the magnitudes of the impedances in 10,000 ppm O₂ in N₂. The gases used were 200 ppm NO₂ in N₂, 200 ppm NO₂ and 10,000 ppm O₂ in N₂, 1000 ppm CO in N₂, and N₂.

Table 6

	MgAl ₂ O ₄	1% ZnMgAl ₂ O ₄	ZnO	WO ₃	NiFe ₂ O ₄	SnO ₂	TiO ₂
NO ₂ in N ₂	0.6245	0.5544	55.85	8.772	5.008	9.243	1.536
NO ₂ in O ₂ /N ₂	0.7680	0.6787	47.38	9.468	12.93	10.56	1.585
CO in N ₂	1.531	1.459	0.1235	0.1865	1.248	0.0051	0.0116
N ₂	0.8242	0.9219	4.1290	1.716	1.327	0.3208	1.055

	MnTiO ₃	NiO	SrNb ₂ O ₆	CeVO ₄	1% Nb:TiO ₂	FeTiO ₃	Pr ₆ O ₁₁
NO ₂ in N ₂	0.8643	0.5692	1.217	0.9847	1.937	1.299	0.5475
NO ₂ in O ₂ /N ₂	0.8475	0.9662	1.228	0.9977	1.674	1.034	0.5452

CO in N ₂	37.35	9.679	0.6501	1.045	0.0112	0.6009	1.184
N ₂	1.264	1.257	1.011	1.001	0.8811	1.028	1.103

	SrTiO ₃	Ba ₂ Cu ₂ O ₅	CuMnFe ₂ O ₄	LaFeO ₃	Zn ₂ V ₂ O ₇
NO ₂ in N ₂	0.6524	0.7869	0.9559	0.8401	1.209
NO ₂ in O ₂ /N ₂	0.7596	0.7834	0.9399	0.8506	1.114
CO in N ₂	0.0178	0.7603	0.6089	2037	0.8529
N ₂	1.061	1.063	1.136	1.756	0.9900

Example 7

- This example illustrates the use of the AC impedance technique for the measurement of the response of 19 metal oxide semiconducting materials in the presence of 4 gas compositions at 550°C. The signals listed in the table are from the AC impedance technique. The signals are the ratios of the magnitudes of the impedances of the materials when exposed to the gas compositions shown to the magnitudes of the impedances in 10,000 ppm O₂ in N₂. The gases used were 200 ppm NO₂ in N₂, 200 ppm NO₂ & 10,000 ppm O₂ in N₂, 1000 ppm CO in N₂, and N₂.

Table 7

	MgAl ₂ O ₄	1% Zn:MgAl ₂ O ₄	ZnO	WO ₃	NiFe ₂ O ₄	SnO ₂
NO ₂ in N ₂	0.9894	0.9583	3.866	2.335	3.025	1.655
NO ₂ in O ₂ /N ₂	0.8937	0.8984	5.272	2.006	3.553	3.390
CO in N ₂	1.046	0.9697	0.0133	0.2034	0.2506	0.0069
N ₂	1.067	1.060	0.7285	0.9526	1.208	0.2666

	TiO ₂	MnTiO ₃	NiO	SrNb ₂ O ₆	CeVO ₄	1% Nb:TiO ₂	FeTiO ₃
NO ₂ in N ₂	1.135	1.010	0.9483	1.006	1.003	1.271	1.193
NO ₂ in O ₂ /N ₂	1.314	1.014	0.5207	1.044	0.9975	1.302	1.073
CO in N ₂	0.0017	44.00	1.194	0.2814	1.104	0.0021	0.6743
N ₂	0.7263	1.280	1.341	0.9830	1.024	0.477	1.054

	Pr ₆ O ₁₁	SrTiO ₃	Ba ₂ Cu ₂ O ₅	CuMnFe ₂ O ₄	LaFeO ₃	Zn ₂ V ₂ O ₇
NO ₂ in N ₂	1.223	0.9055	0.7071	1.148	1.302	1.199
NO ₂ in O ₂ /N ₂	0.9656	0.9881	0.3812	0.9891	0.9429	1.086
CO in N ₂	62.76	0.0029	3.0892	2.557	123.3	0.4726
N ₂	1.495	1.210	1.333	1.681	1.789	0.9034

Example 8

- This example illustrates the use of the AC impedance technique for the measurement of the response of 23 semiconducting materials in the presence of 4 gas compositions at 650-700°C. The signals listed in the table are from the AC impedance technique. The signals are the ratios of the magnitudes of the impedances of the materials when exposed to the gas compositions shown to the magnitudes of the impedances in 10,000 ppm O₂ in N₂. The gases used were 200 ppm NO₂ in N₂, 200 ppm NO₂ & 10,000 ppm O₂ in N₂, 1000 ppm CO in N₂, and N₂.

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Table 8

	MgAl ₂ O ₄	1% Zn:MgAl ₂ O ₄	ZnO	WO ₃	NiFe ₂ O ₄	SnO ₂	TiO ₂
NO ₂ in N ₂	0.9450	1.022	0.4876	0.7151	0.5807	0.5419	0.5617
NO ₂ in O ₂ /N ₂	0.6412	0.8310	1.235	1.281	1.105	0.8265	1.030
CO in N ₂	0.9074	0.9684	0.0348	0.2693	0.0408	0.0238	0.0015
N ₂	1.056	1.100	0.2753	0.6332	0.4421	0.3521	0.3957

	MnTiO ₃	NiO	SrNb ₂ O ₆	CeVO ₄	1% Nb:TiO ₂	FeTiO ₃	Pr ₆ O ₁₁
NO ₂ in N ₂	1.445	1.379	0.8852	1.050	0.5711	0.9072	1.516
NO ₂ in O ₂ /N ₂	0.9561	0.8127	0.9862	1.135	0.8263	0.9524	0.9814
CO in N ₂	113.3	1.782	0.0301	1.565	0.0035	0.4346	8005
N ₂	1.877	1.409	0.8788	1.080	0.2802	0.8050	1.962

	SrTiO ₃	Ba ₂ Cu ₂ O ₅	CuMnFe ₂ O ₄	LaFeO ₃	Zn ₂ V ₂ O ₇
NO ₂ in N ₂	1.051	0.5615	3.401	1.331	0.8631
NO ₂ in O ₂ /N ₂	0.9320	0.9703	1.001	1.013	0.9459